NORTH MAC MIDDLE SCHOOL

8th GRADE PHYSICAL SCIENCE

CURRICULUM GUIDE

Teacher: Richard Cooper

Grade Level: 8th grade

Course: Physical Science

Course Description: This is a hands-on course which teaches 8th grade students the Next Generation Science Standards in physics and chemistry. We will focus on studying concepts related to scientific inquiry, motion and forces, energy and matter, properties of matter, interactions of matter, and waves, electricity, and magnetism. The purpose of 8th Grade Science is to prepare students for high school science curriculum by using strategies aligned with the Common Core State Standards and the Next Generation Science Standards.

Textbook: Title: Physical I Science ISBN: 978-0-07-888004-9 Authors: Glencoe Publisher: McGraw-Hill Companies Publication Date: 2012

Assessment: Number of assessments and point values vary from quarter to quarter and are obtained from assignments, quizzes, chapter tests, labs, and projects. The percent values for assessment are as follows:

Assignments - 25% Tests and quizzes - 50% Projects - 25%

Content – 3 weeks	Assessment	Common Core	Essential Questions
Scientific Problem	Daily Homework	None	What are some steps
Solving	Quizzes		used during science
	Lab/Projects		inquiry?
	Tests		What are the results of
	, ;		scientific inquiry?
		•	What is critical
			thinking?
			Why did scientists
			create the
			International System of
			Units (SI)?
			Why is scientific
			notation a useful tool
			for scientists?
			How can tools, such as
			graduated cylinders

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Content – 4 weeks Chapter 1 – Describing Motion & Chapter 2 – The Laws of Motion	Assessment Daily Homework Quizzes Lab/Projects Tests	Common Core MS-PS2-1. Apply Newton's Third Law to design a solution to a problem involving the motion of two colliding objects. MS-PS2-2. Plan an investigation to provide evidence that the change in an object's motion depends on the sum of the forces on the object and the mass of the object. MS-PS2-4. Construct and present arguments using evidence to support the claim that gravitational interactions are attractive and depend on the masses of interacting objects.	and triple-beam balances, assist physical scientists? Essential Questions How does the description of an object's position depend on a reference points? How can you describe the position of an object in two dimensions? What is the difference between distance and displacement? What is speed? How can you use a distance-time graph to calculate average speed? What are ways velocity can change? What are three ways an object can accelerate? What does a speed-time graph indicate about an object's motion? What are some contact forces and some noncontact forces? What is the law of universal gravitation?
			universal gravitation? How does friction affect the motion of two objects sliding past
	; .		each other? What is Newton's first law of motion? How is motion related to balanced and unbalanced forces? What effect does inertia have on the motion of an object?

Content – 4 weeks	Assessment	Common Core	Essential Questions
Chapter 5 – Energy and	Daily Homework	MS-PS3-2. Develop a	What is energy?
Energy Resources &	Quizzes	model to describe that	What are potential and
Chapter 6 – Thermal	Lab/Projects	when the arrangement	kinetic energy?
Energy	Tests	of objects interacting	How is energy related
		at a distance changes,	to work?
		different amounts of	What are different
		potential energy are	forms of energy?
		stored in the system.	What is the law of
			conservation of energy?
		MS-PS3-3. Apply	How does friction
		scientific principles to	affect energy
		design, construct, and	transformations?
		test a device that	How are different types
		either minimizes or	of energy used?
		maximizes thermal	How are temperature
		energy transfer.	and kinetic energy
		MS-PS3-4. Plan an	related?
		investigation to	How do heat and
	* * * * * * * * * * * * * * * * * * *	determine the	thermal energy differ?
		relationships among	What is the effect of
		the energy transferred,	having a small specific heat?
		the type of matter, the	What happens to a
		mass, and the change	material when it is
		in the average kinetic	heated?
		energy of the particles	In what ways can
		as measured by the	thermal energy be
		temperature of the	transferred?
		sample.	
		111	`
		MS-PS3-5. Construct,	
		use, and present	
		arguments to support	
		the claim that when	
		the kinetic energy of an	
		object changes, energy	
		is transferred to or	÷
		from the object.	
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Content – 4 weeks	Assessment	Common Core	Essential Questions
Chapter 7 –	Daily Homework	MS-PS1-4. Develop a	What is a substance?
Foundations of	Quizzes	model that predicts	How do atoms of
Chemistry & Chapter 8	Lab/Projects	and describes changes	different elements
– States of Matter	Tests	in particle motion,	differ?
		temperature, and state	How do mixtures differ
		of a pure substance	from substance?
		when thermal energy is	How can you classify
		added or removed.	matter?
		added of Tellloved.	What are some
		MS-PS1-6. Undertake a	physical properties of
		design project to	matter?
		construct, test, and	How are physical
		modify a device that	properties used to
	1	either releases or	separate mixtures?
		absorbs thermal	How can a change in
		energy by chemical	energy affect the state
		processes.	of matter?
		processes.	What happens when
		MS-PS3-1. Construct	something dissolves?
		and interpret graphical	What is meant by conservation of mass?
	*	displays of data to	What is a chemical
		describe the	property?
		relationships of kinetic	What are some signs of
		energy to the mass of	chemical change?
		an object and to the	Why are chemical
		speed of an object.	equations useful?
		speed of all object.	What are some factors
			that affect the rate of
		· 4 · · · · ·	chemical reactions?
		· ·	How do particles move
		·	in solids, liquids, and
			gases?
			How are the forces
			between particles
			different in solids,
			liquids, and gases?
			How is temperature
			related to particle
			motion?
		7 - 7 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -	How are temperature
			and thermal energy
			different?
			What happens to thermal energy when
	1	1	I thormal anargy when

	matter changes from one state to another? How does the kinetic molecular theory describe the behavior of a gas? How are temperature, pressure, and volume related in Boyle's law? How is Boyle's law different from Charles's law?
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Assessment	Common Core	Essential Questions
Daily Homework Quizzes Lab/Projects Tests	MS-PS1-1. Develop models to describe the atomic composition of simple molecules and extended structures.	What is an atom? How would you describe the size of an atom? How has the atomic model changed over time? What happens during nuclear decay? How does a neutral atom change when its number of protons, electrons, or neutrons changes? How are elements arranged on the periodic table? What can you learn about elements from the periodic table? What elements are metals? What are the
		properties of metals? Where are nonmetals and metalloids on the periodic table? What are the properties of nonmetals and metalloids?
	Quizzes Lab/Projects	Quizzesmodels to describe theLab/Projectsatomic composition ofTestssimple molecules and

Content – 4 weeks	Assessment	Common Core	Essential Questions
Chapter 11 – Elements	Daily Homework	MS-PS1-2. Analyze and	How is an electron's
and Chemical Bonds &	Quizzes	interpret data on the	energy related to its
Chapter 12 – Chemical	Lab/Projects	properties of	distance from the
Reactions and	Tests	substances before and	nucleus?
Equations	1000	after the substances	Why do atoms gain,
Equations		interact to determine if	lose, or share
		a chemical reaction has	electrons?
		occurred.	How do elements differ
		occurred.	from the compounds
		MS-PS1-5. Develop and	they form?
		use a model to	What are some
		describe how the total	common properties of
		number of atoms does	a covalent compound?
			Why is water a polar
		not change in a	
		chemical reaction and thus mass is conserved.	compound? What is an ionic
		thus mass is conserved.	
			compound? How do metallic bonds
			differ from covalent
		, \$2° e	and ionic bonds?
			What are some signs
			that a chemical
			reaction might have
			occurred?
			What happens to
			atoms during a
			chemical reaction?
			What happens to the
			total mass in a
			chemical reaction?
			How can you recognize
			the type of chemical
			reaction by the number
			or type of reactants
		Take the second of the second	and products?
		·	What are the different
			types of chemical
			reactions?
			Why do chemical
			reactions always
			involve a change in
			energy?
		$\int_{X} T_{\chi}(x,y) dy$	What is the difference
			between an
			endothermic reaction

	and an exothermic reaction? What factors can affect the rate of a chemical reaction?
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Content – 4 weeks	Assessment	Common Core	Essential Questions
Chapter 14 – Carbon	Daily Homework	MS-PS1-3. Gather and	How is carbon unique
Chemistry	Quizzes	make sense of	compared to other
	Lab/Projects	information to describe	elements?
	Tests	that synthetic materials	How does carbon bond
		come from natural	with other carbon
		resources and impact	atoms?
		society.	What are the three
			common functional
			groups of organic
			compounds?
			What are polymers?
			What are biological
			molecules?
			What are some groups
			of carbon compounds
			found in living
			organisms?

Content – 3 weeks	Assessment	Common Core	Essential Questions
Chapter 15 – Waves	Daily Homework	MS-PS4-1. Use	What is a wave?
	Quizzes	mathematical	How do different types
	Lab/Projects	representations to	of waves make
	Tests	describe a simple	particles of matter
		model for waves that	move?
		includes how the	Can waves travel
		amplitude of a wave is	through empty space?
		related to the energy in	What are properties of
		a wave.	waves?
			How are the frequency
		MS-PS4-2. Develop and	and the wavelength of
		use a model to	a wave related?
		describe that waves	What affects wave
		are reflected,	speed?
		absorbed, or	How do waves interact
		transmitted through	with matter?
		various materials.	

	What are reflection,
	refraction, and
	diffraction?
	What is interference?

Content – 2 weeks	Assessment	Common Core	Essential Questions
Chapter 17.3 –	Daily Homework	MS-PS4-3. Integrate	How are different types
Electromagnetic Waves	Quizzes	qualitative scientific	of electromagnetic
	Lab/Projects	and technical	waves used for
	Tests	information to support	communication?
		the claim that digitized	What are some
		signals are a more	everyday applications
		reliable way to encode	of electromagnetic
		and transmit	waves?
		information than	What are some medical
		analog signals:	uses of
			electromagnetic
			waves?

Content – 4 weeks	Assessment	Common Core	Essential Questions
Chapter 19 – Electricity	Daily Homework	MS-PS2-3. Ask	How do electrically
& Chapter 20 –	Quizzes	questions about data	charged objects
Magnetism	Lab/Projects	to determine the	interact?
	Tests	factors that affect the	How can objects
		strength of electric and	become electrically
		magnetic forces.	charged?
			What is an electric
	1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -	MS-PS2-5. Conduct an	discharge?
		investigation and	What is the
		evaluate the	relationship between
		experimental design to	electric charge and
		provide evidence that	electric current?
		fields exist between	What are voltage,
		objects exerting forces	current, and
		on each other even	resistance? How do
		though the objects are	they affect each other?
		not in contact.	What are the basic
			parts of an electric
			circuit?
			How do the two types
			of electric circuits
			differ?
			What types of forces
			do magnets apply to
			other magnets?

Why are some materials magnetic? Why are some magnets temporary while others are permanent? Why does a magnet apply a force on an electric current? How do electromagnets and permanent magnets differ? How do electric motors use magnets? How can a wire and magnet produce an electric current? How do electric generators create an electric current? How are transformers used to bring an electric current into your home?

Next Generation Life Science Standards:

- MS-PS1-1. Develop models to describe the atomic composition of simple molecules and extended structures.
- MS-PS1-2. Analyze and interpret data on the properties of substances before and after the substances interact to determine if a chemical reaction has occurred.
- MS-PS1-3. Gather and make sense of information to describe that synthetic materials come from natural resources and impact society.
- MS-PS1-4. Develop a model that predicts and describes changes in particle motion, temperature, and state of a pure substance when thermal energy is added or removed.
- MS-PS1-5. Develop and use a model to describe how the total number of atoms does not change in a chemical reaction and thus mass is conserved.
- MS-PS1-6. Undertake a design project to construct, test, and modify a device that either releases or absorbs thermal energy by chemical processes.
- MS-PS2-1. Apply Newton's Third Law to design a solution to a problem involving the motion of two colliding objects.

- MS-PS2-2. Plan an investigation to provide evidence that the change in an object's motion depends on the sum of the forces on the object and the mass of the object.
- MS-PS2-3. Ask questions about data to determine the factors that affect the strength of electric and magnetic forces.
- MS-PS2-4. Construct and present arguments using evidence to support the claim that gravitational interactions are attractive and depend on the masses of interacting objects.
- MS-PS2-5. Conduct an investigation and evaluate the experimental design to provide evidence that fields exist between objects exerting forces on each other even though the objects are not in contact.
- MS-PS3-1. Construct and interpret graphical displays of data to describe the relationships of kinetic energy to the mass of an object and to the speed of an object.
- MS-PS3-2. Develop a model to describe that when the arrangement of objects interacting at a distance changes, different amounts of potential energy are stored in the system.
- MS-PS3-3. Apply scientific principles to design, construct, and test a device that either minimizes or maximizes thermal energy transfer.
- MS-PS3-4. Plan an investigation to determine the relationships among the energy transferred, the type of matter, the mass, and the change in the average kinetic energy of the particles as measured by the temperature of the sample.
- MS-PS3-5. Construct, use, and present arguments to support the claim that when the kinetic energy of an object changes, energy is transferred to or from the object.
- MS-PS4-1. Use mathematical representations to describe a simple model for waves that includes how the amplitude of a wave is related to the energy in a wave.
- MS-PS4-2. Develop and use a model to describe that waves are reflected, absorbed, or transmitted through various materials.
- MS-PS4-3. Integrate qualitative scientific and technical information to support the claim that digitized signals are a more reliable way to encode and transmit information than analog signals.